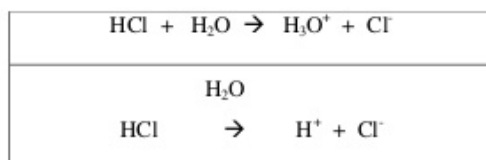




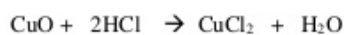
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ACID AND BASES

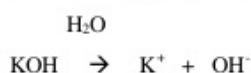
1. An acid is a chemical compound that produce hydrogen ion, H^+ or **hydroxonium, H_3O^+** when dissolves in water.



2. An acid shows its **acidic properties** only in the presence of **water**.
3. A base is defines as a chemical substance that can neutralise an acid to produce a salt and water.
Example:



4. An alkali is defined as a chemical compound that dissolves in water to produce hydroxide ions, OH^- .
Example:



5. Define weak acid :

.....

6. Define strong acid :

.....

7. Define strong alkali :

.....

8. Define weak alkali:

.....

9. Define basicity : the number of ionisable atoms per molecule of an acid.

10. Using a suitable examples explain what is meant by

a) Monoprotic acid :

.....

b) Diprotic acid :

.....

c) Triprotic acid :

.....



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